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Synthesis and Vibrational Spectroscopic Investigation of Methyl L-Prolinate Hydrochloride: A Computational Insight

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KEYWORDS density functional theory, Hartree-Fock, highest occupied molecular orbital, lowest unoccupied molecular orbital, methyl L-prolinate hydrochloride, synthesis, thermodynamic functions

INTRODUCTION

L-Proline is an amino acid, which has one secondary amine and a carboxylic acid functional group with a chiral center. It is not an essential amino acid because the human body can synthesize it. L-Proline is one DNA-encoded amino acid out of 20 amino acids. It is a five-member hetero ring system having one nitrogen and four carbon annular atoms, and this contributes to it having a rigid ring structure, which leads to its special characteristic features such as the bending template effect in peptide chains. The synthetic products of L-proline derivatives show significant biological activity and organo catalytic behavior. Especially, they are used in enantio-selective aldol, Mannich, and Michael addition reactions and in the synthesis of certain macrocyclic molecular systems. [1–5]

L-Proline is the derivative of glutamine amino acid. It has a significant role in the human biological system. It contributes to the healthy functioning of the bone, muscles, skin, and immune system. A deficiency in this amino acid might lead an individual to have tears in tissues and very slow healing ability.

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